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Transition metal dichalcogenides (TMDs) have attracted intensive research interest due to their diverse properties. However, ferromagnetism is not observed in layered TMDs, except for monolayer VSe₂. In this study, we report the synthesis of a bulk ferromagnetic material (LiOH)_{0.1}VS₂ based on topochemical reactions. The results

demonstrate that the (LiOH)_{0.1}VS₂ crystal exhibits strong anisotropic ferromagnetism below a critical temperature of 40 K. Calculations uncover that the in-plane strains in a $\sqrt{3} \times \sqrt{7}$ VS₂ superlattice can induce large magnetic anisotropic energy, which stabilizes the longrange ferromagnetic order. The findings provide a new approach to induce ferromagnetism in bulk TMD materials.

TMDs, as a large family of layered van der Waals (vdW) materials of general formula MX_2 (M = metals, X = chalcogens), have been the subject of intensive research due to their rich properties that make them attractive for both fundamental studies of novel physical phenomena and for practical applications.¹⁻³ With the merits of adjustable structural phase and layer thickness and especially the highly tunable chemical compositions in both the M and X sites, TMDs can display a wide range of electronic properties, such as optoelectronic properties and luminance (e.g., MoS₂, WS₂),⁴⁻⁶ charge density waves and superconductivity (e.g., NbSe₂ and TaS₂),⁷⁻⁹ and valleytronic properties with a controllable valley degree of freedom (e.g., MoS₂, WSe₂).¹⁰ However, this family has been missing one crucial member: ferromagnets.^{11–14} It is only recently that ferromagnetism has been experimentally observed in atomic thick VSe₂ films, and the findings are still debatable. Until now, there has been no experimental success in inducing ferromagnetism in bulk TMD-based materials.

To induce intrinsic ferromagnetism in TMDs, the M site of the MX₂ material must be occupied by high-spin state atoms.¹⁵ The only M element that, in its high spin state, can meet the

Ferromagnetism induced by in-plane strain in a bulk VS₂-based superlattice: (LiOH)_{0.1}VS₂†

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> requirements for chemical stability and the formation of a layered structure in TMDs is vanadium (V). With the 1T structural configuration and high-spin state of M element V, bulk VX_2 (X = S, Se) is experimentally determined to be paramagnetic,^{16,17} but it is theoretically proposed to exhibit ferromagnetic order down to the monolayer limit.¹⁸⁻²⁰ Recent studies by M. Bonilla et al. reported the observation of a ferromagnetic state up to room temperature in monolayer VSe2.21 Later, several similar experiments failed to reproduce the ferromagnetism in monolayer VSe₂, making the findings a matter of debate.²²⁻²⁷ For instance, the results of ARPES on monolayer VSe₂ show that there are no exchange-split electronic bands and ferromagnetically coupled V atoms down to 10 K.²² It is also reported based on the results of STM/nc-AFM that ferromagnetism does exist in the VSe2 monolayer, but only in the region with abundant Se-defects,^{25,26} suggesting that the observed ferromagnetism should not be intrinsic. Theoretically, it is proposed that intrinsic ferromagnetic order can be stabilized in the strained VS₂/VSe₂ monolayer, wherein both the magnetic moments and strength of magnetic coupling increase rapidly with increasing isotropic strain.¹⁹ Given these conflicting experimental and theoretical results, we aim to fabricate a bulk V-based TMD material that is free of abundant defects, wherein intrinsic ferromagnetism may be induced in the strained VS₂/VSe₂ lattice.

> In this study, we report the successful realization of ferromagnetism in a bulk superlattice (LiOH)_{0.1}VS₂, which is constructed based on paramagnetic 1T-VS₂. To ensure the VS₂ layer is free of the defects that generally occurred in samples synthesised at high temperatures, the title compound is obtained through a low temperature topochemical reaction. K_{0.6}VS₂ single crystal, which is known to have defect-free T-VS₂ layers,²⁸ is used as the parent matrix. The structural solution of a (LiOH)_{0.1}VS₂ single crystal and maxim entropy charge density analysis revealed a $\sqrt{3}$ × $\sqrt{7}$ commensurate superlattice in stoichiometric VS₂ layers, suggesting that in-plane strains are imposed to the VS₂ matrix through insertion of hydrate layers. Magnetic properties measurement revealed that below ~ 40 K, the bulk material becomes a ferromagnet with significant magneto-crystalline anisotropy. Extensive

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Fig. 1 (a) Reaction scheme for the manipulation of $(LiOH)_{0.1}VS_2$ and $1T-VS_2$. (b) Powder X-ray diffraction patterns along the (00l) direction of products $(LiOH)_{0.1}VS_2$ and $1T-VS_2$ and the matrix K_xVS_2 single crystals. (c) SEM image and EDS compositional mapping of $(LiOH)_{0.1}VS_2$. (d) XPS pattern of V 2p, S 2p, Li 1s, and O 1s (OH^-) in $(LiOH)_{0.1}VS_2$.

density functional theory (DFT) calculations are also applied to uncover the origin of the emergence of ferromagnetism in stoichiometric VS_2 layers.

The topochemical reaction starting from the $K_{0.6}VS_2$ single crystalline matrix to (LiOH)_{0.1}VS₂ and 1T-VS₂ is demonstrated in Fig. 1a. The as-prepared (LiOH)_{0.1}VS₂ single crystal was characterized by powder X-ray diffraction (PXRD). As shown in Fig. 1b, after 7 days of reaction in the LiOH aqueous solution, the (00l) reflections of the K_{0.6}VS₂ precursors are completely replaced by a new set of ones that systematically shift to lower angles, which indicates the presence of new intercalated species in between the VS_2 layers. The chemical stoichiometry of the starting material K_{0.6}VS₂ is determined by scanning electron microscopy (SEM) and energydispersive X-ray spectroscopy (EDS) mapping (Fig. S1, ESI⁺). The homogeneous distribution of V and S with a molar ratio of 1:2 is also confirmed by EDS in (LiOH)_{0.1}VS₂, as shown in Fig. 1c. X-ray photoelectron spectroscopy (XPS) analysis was used to identify the species and oxidation states of elements such as Li, OH⁻, V, and S, in (LiOH)_{0.1}VS₂. As shown in Fig. 1d, the Li 1s (55 eV) and O 1s (531 eV) peaks appeared in $(LiOH)_{0.1}VS_2$, which are identical to the reported LiOH spectra, suggesting the insertion of LiOH groups.²⁹ Moreover, only one kind of XPS peak of V 2p (517 and 525 eV) and S 2p (159.5 and 160.5 eV) was detected in (LiOH)_{0.1}VS₂, with their peak positions consistent with the stoichiometric 1T-VS₂, suggesting that after reaction the VS2 layer becomes charge neutral and there is basically no V transferred to the interlayer space.³⁰ From this, we can conclude that VS₂ layers are inserted by charge-neutral groups LiOH. The calculated Li:OH-:V:S atomic ratio from XPS data (0.1:0.1:1:2) is in agreement with the results of ICP-AES in Table S1 (ESI[†]), suggesting that the material has good lateral uniformity and that a trace amount of LiOH molecules is incorporated in-between the VS₂ layers.

We then used single-crystal X-ray diffraction (SXRD) to determine the crystal structure of (LiOH)_{0.1}VS₂. The SXRD data can be indexed by a $P\overline{1}$ unit cell, with lattice parameters a = 5.6557(6) Å, b = 8.6279(2) Å and c = 9.0551(7) Å. To probe the light elements of Li, O and H, high precision electron density distributions (EDD) of $(LiOH)_{0.1}VS_2$ (wR_F = 0.021) were constructed based on the model independent maximum-entropy method (MEM).³¹ It's clearly shown that in-between the VS₂ layers, there are two sites with charge aggregation (Fig. 2a). Considering that the XPS data suggest no noticeable interlayer V ions, it is reasonable to ascribe the heavier charge aggregation to O and the lighter one to Li. Subsequent refinement on the occupancy of O & Li leads to a value close to the formula determined by chemical analysis. As shown in Fig. 2b, the interlayer LiOH exists in molecular form with straight rod-like shape, similar to the previous findings.^{32,33} Intriguingly, the VS₂ layer in (LiOH)_{0.1}VS₂ forms a $\sqrt{3} \times$ $\sqrt{7}$ ($\sqrt{3} \times 3.26 = 5.65$, $\sqrt{7} \times 3.26 = 8.63$) super cell in comparison with hexagonal 1T-VS₂. The V-V bond length (3.260 Å) is approximately 1.2% longer than bulk 1T-VS₂, suggesting that the insertion of LiOH molecules applies $\sim 1.2\%$ tensile strain to the host layer. The [VS₆] polyhedron is also distorted, resulting in three inequivalent V-S bond lengths (Fig. 2c). The large anisotropic in-plane strain and a decrease in crystal symmetry caused by the insertion of molecules are also observed in some FeSe intercalated compounds.34 The refined crystal structural data is shown in Tables S2 (ESI⁺), and the bond lengths and angles for $(LiOH)_{0,1}VS_2$ are deposited in Table S3 (ESI[†]).

Magnetic properties of the $(\text{LiOH})_{0.1}\text{VS}_2$ single crystals were investigated using a PPMS-VSM system. Fig. 3a and b show the magnetic moment against the magnetic field (*M*-*H*) curves measured at different temperatures with the field parallel and perpendicular to the crystal. The *M*-*H* hysteresis loops observed



Fig. 2 (a) Charge density map for (LiOH)_{0.1}VS_2. (b) Structure model for (LiOH)_{0.1}VS_2. (c) Scheme for the $\sqrt{3}\times\sqrt{7}$ superlattice in (LiOH)_{0.1}VS_2.

at 10 K well demonstrate the ferromagnetism in $(\text{LiOH})_{0.1}\text{VS}_2$, while both the as-grown 1T-VS₂ and K_{0.6}VS₂ are paramagnetic (Fig. S2 and S3, ESI[†]). Notably, the in-plane hysteresis loop is significantly larger than the out-of-plane one, indicating the strong anisotropy of ferromagnetism. The easy axis in $(\text{LiOH})_{0.1}\text{VS}_2$ is parallel to the *xy* plane, which is in line with the findings in the VSe₂ monolayer.²¹ Moreover, the small coercivity field (200 Oe) suggests that $(\text{LiOH})_{0.1}\text{VS}_2$ is a soft magnet material, which is also similar to monolayer VSe₂.²¹ Fig. 3c shows the zero-field cooling (ZFC) and field cooling (FC) curves with in-plane and out-of-plane fields. The magnetic moments decrease with increasing temperature and both vanish at around 40 K. This behavior again verifies the existence of ferromagnetic order in $(\text{LiOH})_{0.1}\text{VS}_2$. Fig. 3d shows the corresponding temperature-dependent inverse *M*–*T* curve. Obviously, the ferromagnetic–paramagnetic (FM–PM) transition occurs at a critical temperature (T_c) of about 40 K. We found that both the in-plane and out-of-plane high-temperature (>50 K) data can be well fitted by Curie–Weiss's law,

$$\chi(T) = \frac{C}{T - \theta}$$

where θ and *C* are the Weiss and the Curie constants. The fitting yields a θ of 140 K for in-plane and 56 K for out-of-plane fields, respectively. Such a large difference in θ suggests the anisotropic paramagnetic behavior of the (LiOH)_{0.1}VS₂. This behavior, along with the anisotropic *M*–*H* data in the ferromagnetic state, confirms the existence of strong magneto-crystalline anisotropy.

To gain an in-depth understanding, we performed DFT calculations to reveal the origin of the ferromagnetic orders in (LiOH)_{0.1}VS₂. We first compared the magnetic ground states of (LiOH)_{0.1}VS₂ and bulk 1T-VS₂ and found that both have ferromagnetic ground states (as shown in Fig. S4 and S5, ESI[†]). For longrange magnetism in 2D systems, the magnetic anisotropy energy (MAE) plays an important role in stabilizing the ferromagnetism at finite temperatures.^{35–37} Fig. 4a shows the angular dependent MAE, which is evaluated by rotating the spin of V atoms in the xy and xz planes. The MAEs show that MAE of both $(LiOH)_{0.1}VS_2$ and 1T-VS₂ are nearly isotropic in the xy plane, but are anisotropic in the xz plane. The easy axes of both materials are in the xy plane. Moreover, the MAE of $(LiOH)_{0.1}VS_2$ (91.88 µeV) is more than three times that of 1T-VS₂ (26.94 µeV), resulting in the long-range ferromagnetism of $(LiOH)_{0.1}VS_2$, which is absent in 1T-VS₂. To further elucidate the origin of the enhancement of MAE in $(LiOH)_{0.1}VS_2$, a set of trial structures are built (Fig. S6, ESI⁺).



Fig. 3 (a) *M*–*H* loops obtained with an in-plane magnetic field at 10 K and 200 K. (b) *M*–*H* loops obtained with an out-of-plane magnetic field at 10 K and 200 K. (c) The field cooling (FC) and zero-field cooling (ZFC) curves obtained with an in-plane and an out-of-plane magnetic field of 1 T. (d) *H*/*M versus* temperature curves, showing the ferromagnetic transition at $T_c \approx 40$ K.



Fig. 4 Angular dependence of magnetic anisotropic energy (MAE) in (a) xz and (b) xy planes. The definition of the structures can be found in Fig. S6 (ESI†).

Compared with 1T-VS₂, (LiOH)_{0.1}VS₂ can be considered as 3R stacking of the $\sqrt{3} \times \sqrt{7}$ superlattice of T-VS₂ layers with larger in-plane lattice constant and interlayer distance (Fig. 4b). Thus, four aspects possibly induce the enhancement of MAE in (LiOH)_{0.1}VS₂: in-plane strain, in-plane distortion, interlayer distance, and stacking ordering. Based on 1T-VS2, three artificial structures are subsequently constructed, as illustrated in Fig. S6 (ESI \dagger). Compared with (LiOH)_{0.1}VS₂, VS₂-d has the same interlayer distance, VS₂-d-strain has the same interlayer distance and lattice constant, while $3R-(LiOH)_{0.1}VS_2$ has the same interlayer distance, lattice constant and stacking ordering. By analyzing their MAEs (Fig. 4), we find that only in-plane tensile strain (the orange line in Fig. 4a) significantly enhances the MAE of VS₂, and in-plane distortion has only a minor influence on MAE, while stacking ordering and interlayer spacing have negligible impacts. Beside the in-plane tensile strain, its also proposed that a large concentration of Se defects may also induce the magnetism in monolayer VSe₂.^{25,26} Considering that in (LiOH)_{0.1}VS₂, both the chemical analysis data and diffraction analysis have precluded the existence of abundant S vacancies within the resolution limits, the in-plane strain induced magnetism is still the most promising scenario. Moreover, we believe that when we reduce $(LiOH)_{0,1}VS_2$ to a few layers or even a monolayer, it can achieve greater flexibility in controlling the in-plane strain of VS₂ by manipulating factors such as substrate type and defects. This, in turn, holds the potential to realize higher $T_{\rm c}$ two-dimensional ferromagnetism.

In summary, bulk single crystals of $(\text{LiOH})_{0.1}\text{VS}_2$ have been successfully synthesized using the topochemical method, exhibiting intrinsic ferromagnetic ordering with a T_c of ~ 40 K. The occurrence of ferromagnetic order in this $\sqrt{3} \times \sqrt{7}$ VS₂ superlattice provides strong evidence that applying in-plane strain in VS₂/VSe₂-based materials can introduce ferromagnetism. The findings reported here are expected to open the gate to more TMD-based ferromagnets, and will contribute to a better understanding of the emergence of ferromagnetism in 2D TMD materials as well.

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Conflicts of interest

There are no conflicts to declare.

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